

Gene Cloning, Overproduction, and Characterization of Thermolabile Alkaline Phosphatase from a Psychrotrophic Bacterium

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The gene encoding alkaline phosphatase from the psychrotrophic bacterium *Shewanella* sp. SIB1 was cloned, sequenced, and overexpressed in *Escherichia coli*. The recombinant protein was purified and its enzymatic properties were compared with those of *E. coli* alkaline phosphatase (APase), which shows an amino acid sequence identity of 37%. The optimum temperature of SIB1 APase was 50 °C, lower than that of *E. coli* APase by 30 °C. The specific activity of SIB1 APase at 50 °C was 3.1 fold higher than that of *E. coli* APase at 80 °C. SIB1 APase lost activity with a half-life of 3.9 min at 70 °C, whereas *E. coli* APase lost activity with a half-life of >6 h even at 80 °C. Thus SIB1 APase is well adapted to low temperatures. Comparison of the amino acid sequences of SIB1 and *E. coli* APases suggests that decreases in electrostatic interactions and number of disulfide bonds are responsible for the cold-adaptation of SIB1 APase.

Key words: alkaline phosphatase; psychrotrophic bacterium; *Shewanella* sp.; cold-adaptation; thermal stability

Microorganisms that can grow at low temperatures, such as psychrophiles and psychrotrophs, usually produce cold-adapted enzymes. A cold-adapted enzyme has been specified by an increase in catalytic efficiency at low temperatures, a downward shift in the apparent optimum temperature for activity, and a reduction in stability at moderate temperatures, as compared to its mesophilic counterpart.^{1–6} Because these properties are useful in various applications,^{7,8} understanding of the structural determinants of the cold-adaptation is important not only to answer questions as to how these enzymes adapt to cold environments, but also to develop

a technique to engineer cold-adapted enzymes in a rational manner. The molecular basis of cold-adaptation, however, remains largely unknown. One promising strategy to analyze the adaptation mechanisms of enzymes to cold environments is to compare the structure and function of a given enzyme from psychrophiles or psychrotrophs with those of a mesophilic counterpart and identify the amino acid substitutions responsible for these differences.

Alkaline phosphatase (APase) (EC 3.1.3.1) is a non-specific phosphomonoesterase that is widely distributed in various organisms from bacteria to humans.⁹ Of these APases, *E. coli* APase has been the most extensively studied for structures and functions.^{10–16} This enzyme is located in the periplasmic space as a homodimer and is involved in the acquisition of phosphate from esters when free inorganic phosphate is depleted. Each subunit is composed of 449 amino acid residues and contains two Zn²⁺ and one Mg²⁺ ions. The crystal structure of this enzyme has been determined, and the amino acid residues involved in the catalytic function, metal binding, and substrate binding have been identified. Hence, a system using *E. coli* APase as one of a pair of mesophilic and psychrophilic/psychrotrophic APases has potential for exploring the molecular mechanisms of cold-adaptation in proteins.

The genes encoding APases from the Antarctic psychrophile TAB5 (TAB5 APase),¹⁷ the marine psychrophile *Vibrio* sp. G15-21 (*Vibrio* APase),¹⁸ and the psychrophile *Shewanella* sp. (SCAPase)¹⁹ were cloned and overexpressed in *E. coli*. These proteins show amino acid sequence identities of 32–34% to *E. coli* APase but these proteins may not be suitable as a partner of *E. coli* APase to analyze the cold-adaptation mechanism of enzymes for the following reasons. The

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Abbreviations: APase, alkaline phosphatase; pNPP, *p*-nitrophenyl phosphate; pNP, *p*-nitrophenol

recombinant protein of TAB5 APase shows typical features of a cold enzyme, but it is relatively small in size (353 amino acids) and apparently exists in a multimeric form¹⁷⁾ instead of a dimeric one. Likewise, the recombinant protein of *Vibrio* APase is very heat labile, but it apparently exists in a monomeric form.¹⁸⁾ The recombinant protein of SCAPase did not show typical features of a cold enzyme. Its optimal temperature for activity is 70 °C,¹⁹⁾ which is shifted upward by 30 °C compared to that of the native enzyme.²⁰⁾

Shewanella sp. strain SIB1 is a psychrotrophic bacterium that grows most rapidly at 20 °C.²¹⁾ This strain can grow even at 0 °C, but cannot grow at temperatures exceeding 30 °C. Ribonuclease HI²²⁾ and FKBP22 with peptidyl prolyl *cis-trans* isomerase activity²³⁾ from this strain have been shown to exhibit enzymatic properties characteristic of cold-adapted enzymes. In this study, we cloned the gene encoding APase from this strain (SIB1 APase), overexpressed it in *E. coli*, purified the recombinant protein, and compared its enzymatic activity and stability with those of *E. coli* APase. Based on these results, we discuss the cold-adaptation mechanism of SIB1 APase.

Materials and Methods

Cells and plasmids. The psychrotrophic bacterium *Shewanella* sp. SIB1 was isolated from Japanese oil reservoir water in our laboratory.²¹⁾ *E. coli* AC109 with the *phoA* mutation²⁴⁾ was kindly donated by A. Nishimura. Plasmid pET-25b was from Novagen (Madison, WI, U.S.A.). *E. coli* AC109(DE3) was constructed by lysogenizing *E. coli* AC109 with λ DE3, a recombinant phage carrying the cloned gene for T7 RNA polymerase under *lacUV5* control, using a λ DE3 Lysogenization Kit (Novagen). λ DE3 lysogens can be used to express the target genes cloned into pET vectors. The *E. coli* transformants were grown in NZCYM medium (Novagen) containing 50 mg/l ampicillin.

Materials. *E. coli* APase and *p*-nitrophenyl phosphate (pNPP) were obtained from Wako Pure Chemical (Osaka, Japan).

Cloning of the SIB1 APase gene. The genomic DNA of *Shewanella* sp. SIB1 was prepared as previously described²⁵⁾ and used as a template to amplify part of the SIB1 APase gene by PCR. The sequences of the PCR primers were 5'-AGTGGTTATGTTACCGACTCGGC-CGC-3' for the 5'-primer and 5'-GTATGACCGCCG-TTGCCAACC-3' for the 3'-primer. PCR was performed with a GeneAmp PCR system 2400 (Perkin-Elmer Japan, Tokyo) using a KOD polymerase (Toyobo, Kyoto, Japan) according to the procedures recommended by the supplier. The amplified DNA fragment was used as a probe for Southern blotting and colony hybridization to clone the entire SIB1 APase gene. Southern blotting and colony hybridization were carried out using the AlkPhos

Direct system (Amersham Biosciences, Piscataway, NJ, U.S.A.) according to the procedures recommended by the supplier. The DNA sequence was determined with a Prism 310 DNA sequencer (Perkin-Elmer Japan). Nucleotide and amino acid sequence analyses, including the localization of open reading frames, determination of molecular weight, and multiple alignment, were performed using DNASIS software (Hitachi Software, Tokyo). All DNA oligomers for PCR were synthesized by Hokkaido System Science (Sapporo, Japan).

Plasmid construction. Plasmid pET-SAP for overproduction of SIB1 APase was constructed by ligating the DNA fragment, which was amplified by PCR using the cloned DNA fragment containing the SIB1 APase gene as a template, into the *Xba*I-*Sal*I sites of pET-25b. The sequences of the PCR primers were 5'-CCTCTA-GAAAGAAGGAGATATACATATGCTTATGGATGCCATCG-3' for the 5'-primer and 5'-CAACCGTCG-ACGTTGATAGCGGAGG-3' for the 3'-primer, where underlined bases show the positions of the *Xba*I (5'-primer) and *Sal*I (3'-primer) sites. In this plasmid, the initiation and termination codons of the SIB1 APase gene are located 70-bp downstream of the *Xba*I site and 40-bp upstream of the *Sal*I site respectively.

Overproduction and purification. *E. coli* AC109(DE3) was transformed with pET-SAP, in which the transcription of the SIB1 APase gene is controlled by the T7 promoter, and grown at 37 °C. When the absorbance at 660 nm of the culture reached 0.5–1.0, 1 mM of isopropyl β -D-thiogalactopyranoside (IPTG) was added to the culture medium and cultivation was continued at 15 °C for 15–20 h. Cells were then harvested by centrifugation at 4000 *g* for 7 min. Recombinant proteins localized in the periplasmic space of the *E. coli* cells were released into the external medium by osmotic shock²⁶⁾ and fractionated by ammonium sulfate precipitation. The protein precipitated between 20 and 40% ammonium sulfate was dissolved in 50 mM Tris-HCl (pH 8.0) containing 5 mM MgCl₂ and applied to a Hitrap Q column (Amersham Biosciences) equilibrated with the same buffer. The protein was eluted from the column with a linear gradient of 0–0.5 M NaCl. The fractions containing highly purified proteins were combined and used for further characterization. The purity of the protein was analyzed by SDS-PAGE,²⁷⁾ followed by staining with Coomassie Brilliant Blue R250.

Protein concentration. The protein concentration was determined from UV absorption using an A₂₈₀ value of 1.0 and 0.70 for 0.1% solutions of the mature forms of SIB1 APase (residues 28–455) and *E. coli* APase respectively. These values were calculated using $\epsilon = 1576 \text{ M}^{-1}$ for Tyr and 5225 M^{-1} for Trp at 280 nm.²⁸⁾

Biochemical characterizations. The molecular mass of the protein was estimated using a column (1.6 ×

						$\alpha 1$	$\beta 1$		
SIB1	1	-----	--MHFLSTYQ	RIISRLLLVS	SLIVTGMAD	EVIMPPAVAN	TASNNVPTVG	PSRFKNITIM	IGDGMG-PAY
SCAP	1	-----	-----	-----	-----	-----	-----	-----	IGDGMG-PAY
TAB5	1	-----	-----	-M	KLKKIVFTLI	ALGLFSCKTT	SVLVKNEPQL	KTP-----	-----
MR-1	1	-----	-----	MSI	TKAPILLLGM	SLWLSSPTFA	-----	AEAGAN	ETD---PVKA
Eco	1	MKQSTIALAL	LPLLETPVTK	ARTPEMPVLE	NRAAQGDITA	PGGARRLTGD	QTAALRDSLS	DKPAKNIHIL	IGDGMG-DSE
Bsu3	1	-----	-----	-----	-----	-----	-----	-----	IGDGMG-VSY
TFD	1	-----	-----	-----	-----	-----	-----	-----	VYDGFSWEDY
Tma	1	-----	-----	-----	-----	-----	-----	-----	IGDGMG-LSQ
			$\alpha 2$	$\alpha 3$	$\beta 2$	$\alpha 4$			
SIB1	68	TIAYRYKDN	PDTEE----	-TEQVFDRL	LVGMASTYPA	SVSCYVTDSDA	ASATALSTGV	KSYNGAIAVD	TEKRPLTILM
SCAP	14	TSAYRYFODN	PDTEE----	-TEQVFDRL	LVGMASTYPA	RESCYVTDSDA	ASATALATCF	KSYNGAIAVD	INRKPLTILM
TAB5	50	ISSTFYFKEG	T-----	-PNYHCFKNI	GLIKTSSRE	-----	LVTDSDA	SCATAPSSCGI	KTYNNAICVA
MR-1	56	TSAYRYKDN	RDTEE----	-TEQVFDRL	LVGMASTYPA	SVSCYVTDSDA	AAATAIATGV	KSYNGAIAVD	TQKQPLTILM
Eco	80	ITAAARNYAE	AGGFFKGIDA	LPLTGQYTHY	ALNKKTGKED	-----	YVTDSDA	ASATAWSTGV	KTYNCAICVD
Bsu3	59	TSAYRYKDN	KKIKV----	-VEPTAFDQY	LVCQQTPTED	DPEQNVTDSDA	AAATAMSGCI	KTYNNAIAVD	NDGSEAKIVL
TFD	55	ATAQAMARR	QGRVL----	-ALERLLARY	PNGLIIMYSI	TS--VVTSS	AAGNAPSCGI	KTVNGGLAIH	AGGTPLKPF
Tma	37	VYLTSMLEG	-----	-----	-----	NS--VVTSDA	AAGTALASCF	KTNNGMNINL	PGTVPVPTIF
			$\alpha 5$	$\beta 3$	$\alpha 6$	$\alpha 7$	$\alpha 8$	$\beta 4$	
SIB1	142	EKAKALGLST	GVAVTSCVNE	ATPAAFLAHN	ESRS-----	-----	NYIDIAE	AYITSDA	-----
SCAP	88	OMAKARCMST	GVAVTSCVNE	ATPAAFLAHN	ESRK-----	-----	NYEALAA	DMIKSDA	-----
TAB5	116	EITAAALNNIK	GVAVTSSITH	ATPAAFLAHN	INRG-----	-----	LEEETAM	DMTESDL	-----
MR-1	130	EKAKALGLST	GVAVTSCVNE	ATPAAFLAHN	ESRK-----	-----	NYDALAL	SYLQINA	-----
Eco	156	EAKAAAGLAT	GNVSTAELOD	ATPAAFLAHN	TSRRCYGPSA	TSEKCPGNAL	EKGKGSITE	QLINARA	-----
Bsu3	133	EPAKEKCKAT	GVAVTSSITH	ATPAAFLAHN	HSRK-----	-----	NMNSTAD	DYFDEMVGK	HKIDVILGGC
TFD	127	AAAKEACKAV	GVAVTSSITH	ATPAAFLAHN	PDRN-----	-----	AEERTAE	QYIEFGD	-----
Tma	103	EVAKTYCVRT	GVAVTSSITH	ATPAAFLAHN	KSRD-----	-----	EENELAR	QLVENET	-----
			$\alpha 9$	$\alpha 10$	$\beta 5$	$\alpha 11$	$\beta 6$	$\alpha 12$	
SIB1	197	QQYFTE----	-----	TLL	EQSAKGYCH	I SEFSEIASV	T--TPK--VI	GLPADVQLE	-----
SCAP	143	RRYFSE----	-----	ATV	SOBSAKGYCH	ITELACIDSI	T--QPK--VI	GLPAEVQLE	-----
TAB5	175	LNVTFRKRD	-----	KKD	VLAIIKCNCF	TINTTGTDF	SSIASNRKMG	FLIADKAMF	-----
MR-1	185	QKVFSE----	-----	ELL	AOBTSKGYCH	ITRFEEIASI	S--QPK--VI	GLPAEVQLE	-----
Eco	230	AKTFAETATA	GEWQK-TLR	EQACAFGYCL	VSDAASINSV	TEANQQKPII	GLPADGNMIV	RWLGPKATYH	GNIDKPAVTC
Bsu3	194	KSNEDRKRDR	-----	NLI	KPEKKAQVSY	VDRDKMLKN	K--DSQ--VI	GLPADGGLF	-----
TFD	182	DRFENFARR	DGK----	DLY	AAFAAKGYGV	VRTPEELARS	N--ATR--LI	GLPADGHVE	-----
Tma	159	WANLEKLDLG	GKRDDNLNLI	ELAKPKGYIV	VKTREELSKI	SA-DSDK--II	ALFAPSHLD	-----	-----
			$\alpha 13$	$\beta 7$	$\alpha 14$	$\alpha 15$	$\beta 8$		
SIB1	241	--WAINDKQA	HKLSHMTOKA	IDLKSCNECC	FVILVEGSLI	DWAGHSNDIA	TAMCENDEHA	SAIEVVEQIV	RQSKDTLIVV
SCAP	187	--WVIDDDTA	NILSKITQKS	IDLKSCNECC	FVILVEGSLI	DWAGHSNDIA	TAMCENDEHA	SAIEVVEQIV	ROHPDTLIVV
TAB5	227	---TMEKGRG	NFLSAAIDLA	IOHLSKDNSA	FFILVEGSLI	DWAGHSNDIA	YLSLSEINDFD	DAIGTALAF	KRDGNTLIVV
MR-1	229	---WALDEKA	KKLTLTOKA	IDLKSCNECC	FVILVEGSLI	DWAGHSNDIA	AAMCENDEHA	NAIEVVEQIV	RQSKDTLIVV
Eco	309	TPNFQRNDV	EITLQMTDKA	IDLKSCNECC	FFILVEGSLI	DKODHAANFC	GOIGETVDLD	BAVQRALEFA	KREGTTLIVV
Bsu3	241	--KKIDRTKDI	ESIKDMINTA	IKKLNKDKDC	FFILVEGSLI	DWAGHSNDIA	GAMSEMEDHE	QAYKATIDFA	KRDGNTLIVV
TFD	235	IDRRFQGLGV	ESIKEMVCAV	IPRIAAHRGC	FVILVEGSLI	DEANHLNDAG	ATLWDVLAAD	EVLILLTAFV	DRNPDTLIVV
Tma	216	--PASSRKEQ	EMLYEMVEKA	LETLKSKDEP	FFILVEGSLI	DREAHNDIIV	GVNKEVVDFT	KAVQVALDEFA	LRGDTLIVV
			$\alpha 13$	$\beta 7$	$\alpha 14$	$\alpha 15$	$\beta 8$		
SIB1	319	TADHNTGGGL	V-GAHDKM--	EWHEPEVL	HKVKASFLVI	ATRAIADNW	QPLTATILGF	TPNEAQYSQI	QSARMQGNP
SCAP	265	TADHNTGGGL	I-GANGFY--	QDDTKLP	KGISASPASI	ATHAIAEDW	QAGVNQQLGF	DVNSTELQQL	TNARMQGNP
TAB5	299	TSDEHTGGCT	L-PAK----	-----	-----	-----	-----	-----	-----
MR-1	307	TADHNTGGGL	I-GADGRN--	NANPEIL	RNISASADTL	AQAASCGDAW	QADLSRGLGF	ALTEEETSKEI	SAARTQGAET
Eco	389	TADHNTGGGL	V-APDT----	-----	-----	-----	-----	-----	-----
Bsu3	320	TADHNTGGGL	I-GADSIY--	NMFSEPI	KAKRRTDFDM	AEKIA-DCAD	VEKTLKTYID	QKILALTKAE	IQSVVEEAAA
TFD	315	VSDEHTGGCT	LYGAGRSYLE	SSVICDLLGA	QKASFEYMR	VLGSAPDAQ	VKEAYQTKG	-----	VSLTDEE
Tma	294	TADHNTGGGL	GLSSGDY--	RVDVDKI	RNFRTKTDWI	MANYSPRE	KFKKATEEYV	-----	GL
			$\alpha 13$	$\beta 7$	$\alpha 14$	$\alpha 15$	$\beta 8$		
SIB1	393	LSIALRKLID	-----	-----	IOSNTGWISG	GHTAMDVQVF	AGGPGARIE	SGHQDNIDIA	IKMFSLLPQS
SCAP	339	LEVALKKIID	-----	-----	TRSYTGWTS	GHTAMDVQVF	AGGPAADIE	KCNQDNIDIA	EKMMSLLPKV
TAB5	325	TEIGPTFSTG	-----	-----	TEIGPTFSTG	GHSATLIEVF	AYGPGSEEF	IGIYENNEIF	HKILKVTKWN
MR-1	381	LAVAIRVVID	-----	-----	KFTDITGWTG	GHTAMDVQVF	AGGPAADIE	NHQDNIDIA	NKIFSLPKP
Eco	434	-----	-----	-----	HTGSLRIA	AYGPHANV	VGLTDCIDLF	YTKAALGLK	-----
Bsu3	391	-KSEVELDID	NAIENIFN--	-----	KRSHTGWTTG	GHTAMDVQVF	AYGPGSEEF	ACQIDNIDIA	KNVFKALQYN
TFD	392	RVYWDPAVRQ	GIQENTMAW	AMVQKNASKP	DRPNIGWSSG	GHTAMFVILL	LYGCLRFVQ	IGLVNDIHVF	RLMGALNLR
Tma	359	SMSKNPKIEL	GRILGKVV--	-----	SVGWTIT	THSCTFPIIE	AFGPGADNE	TCFTDNEFLP	RIIMKLTGYS
			$\alpha 13$	$\beta 7$	$\alpha 14$	$\alpha 15$	$\beta 8$		
SIB1	452	VQTP							
SCAP	398	N							
TAB5	375	Q							
MR-1	440	KPKPTAPPTT	VQPQ						
Bsu3	457	IKINDK							
TFD	472	YQNPVMESEE	ALEILKARPO	GMRHPEDVWA					
Tma	423	LQYPLKKEPV	TK						

Fig. 1. Alignment of the Bacterial APase Sequences.

The most conserved amino acid residue at each position is highlighted in black. Gaps are denoted by dashes. The serine and arginine residues forming the catalytic site (phosphorylation site) and phosphate binding site of *E. coli* APase are denoted by \oplus and \triangle respectively. The amino acid residues forming the three metal binding sites of *E. coli* APase are denoted by asterisks (*). Numbers along the sequences indicate the positions of the amino acid residues which start from the initiator methionine for each protein. The ranges of the α -helices and β -strands of *E. coli* APase are indicated above the sequences. The accession numbers of these sequences are AB073982 for *Shewanella* sp. APase (SCAP), Y18016 for Antarctic strain TAB5 APase (TAB5), AE015527 for *Shewanella* sp. MR-1 APase (MR-1), M29670 for *E. coli* APase (Eco), D88802 for *B. subtilis* APase III (Bsu3), AF079878 for *Thermus* sp. FD3041 APase (TFD), and AE001701 for *T. maritima* APase (Tma). Underlined sequences represent the N-terminal signal sequences, which were either experimentally determined or are estimated using the SignalP V2.0 World Web Server. The signal sequences have been determined for SIB1 APase (this study), TAB5 APase,¹⁷⁾ *E. coli* APase,³¹⁾ and *B. subtilis* APase III.³²⁾

60 cm) of Superdex 200 (Amersham) equilibrated with 20 mM Tris-HCl (pH 8.0) containing 150 mM NaCl. The standard proteins contained in a marker kit (Amersham) were aldolase (158 kDa), bovine serum albumin (67 kDa), ovalbumin (43 kDa), chymotrypsinogen A (25 kDa), and RNase A (14 kDa).

The far-UV (200–260 nm) and near-UV (250–320 nm) CD spectra were measured on a J-725 spectropolarimeter (Japan Spectroscopic, Tokyo, Japan) at 25 °C in 100 mM Tris-HCl (pH 8.0). The protein concentration was 0.18–0.27 mg/ml and 1.2–1.6 mg/ml, and the optical path length of the cell was 2 mm and 10 mm, for the measurement of far-UV and near-UV CD spectra respectively. The mean residue ellipticity (θ , deg cm² dmol⁻¹) was calculated using an average amino acid molecular weight of 110.

The NH₂-terminal amino acid sequence was determined using a Procise 491 automated protein sequencer (Perkin-Elmer Japan).

Enzymatic activity. The APase activity was determined using *p*-nitrophenyl phosphate (*p*NPP) as a substrate. The enzymatic reaction was carried out for 10 min in 50 mM Gly-NaOH (pH 10.5) containing 20 mM MgCl₂, 5 mM ZnCl₂, 100 mM KCl, and 5 mM *p*NPP at 20 °C for SIB1 APase, or in 0.1 M CAPS-NaOH (pH 10.0) containing 0.4 M NaCl, 10 mM MgCl₂, and 5 mM *p*NPP at 60 °C for *E. coli* APase, unless otherwise described. The buffer for *E. coli* APase was identical to that reported previously.²⁹⁾ The reaction was initiated by the addition of the enzyme and terminated by the addition of 0.2 M NaOH. The amount of *p*-nitrophenol (*p*NP) produced by the reaction was determined from the molar absorption coefficient value of 16,200 M⁻¹cm⁻¹ at 410 nm. One unit of enzymatic activity was defined as the amount of the enzyme that produced 1 μ mol *p*NP per min. The specific activity was defined as the enzymatic activity per milligram of protein. Dependencies of the SIB1 APase activity on pH, divalent cation, and salt were analyzed in a condition in which other conditions are optimal for this activity.

Stability against heat inactivation. Stability against heat inactivation was analyzed by incubating the enzyme (0.05–0.1 mg/ml) in 50 mM Tris-HCl (pH 8.0)

containing 5 mM MgCl₂ at various temperatures. At appropriate intervals, aliquots were withdrawn and the enzymatic activity was determined at 20 °C for SIB1 APase and 60 °C for *E. coli* APase.

Results

Cloning

When the amino acid sequences of various bacterial APases are compared with one another, the sequences around the active-site residues are well conserved.^{17–19)} Previously, we reported that RNase HI²²⁾ and FKBP22²³⁾ from *Shewanella* sp. SIB1 show the highest amino acid sequence identities, of 78% and 85% respectively, to those from *Shewanella oneidensis* MR-1. These results suggest that SIB1 APase shows a high amino acid sequence identity to MR-1 APase as well. Therefore, we constructed the PCR primers based on the sequences encoding Ser⁹²-Ala¹⁰⁰ and Gly³⁹⁶-Thr⁴⁰³ of MR-1 APase (accession no. AE015527), and used them to amplify part of the SIB1 APase gene. These regions contain the phosphorylation site and one of the metal binding sites (Ser¹²⁴ and His⁴³⁴ for *E. coli* APase respectively). A PCR using the genomic DNA of *Shewanella* sp. SIB1 as a template produced only a 936-bp DNA fragment, which encodes part of the SIB1 APase sequence. Southern blotting and colony hybridization using this DNA fragment as a probe indicated that a 2.5-kbp *Kpn*I fragment of the SIB1 genome contained the entire SIB1 APase gene (data not shown). Determination of the nucleotide sequence of the SIB1 APase gene indicated that SIB1 APase is composed of 455 amino acid residues with a calculated molecular weight of 49,104 and an isoelectric point (*pI*) of 4.9. A potential Shine Dalgarno (SD) sequence (AGGA) is located seven nucleotides upstream of the initiation codon for translation, and a possible promoter sequence is located upstream of this SD sequence. The nucleotide sequence of the SIB1 APase gene is deposited in DDBJ under accession no. AB190867.

Amino acid sequence

The amino acid sequence of SIB1 APase is compared with those of various APases from psychrophiles, mesophiles, and thermophiles in Fig. 1. SIB1 APase

shows the highest amino acid sequence identity, of 66%, to MR-1 APase.³⁰ It shows amino acid sequence identities of 62% to SCAPase,¹⁹ 33% to TAB5 APase,¹⁷ 37% to *E. coli* APase,³¹ 42% to *B. subtilis* APase III,³² 35% to *Thermus* sp. FD3041 APase (unpublished, accession no. AF079878), and 38% to *Thermotoga maritima* APase.³³ The residues forming the active site of *E. coli* APase¹⁰ are fully conserved in the SIB1 APase sequence, except for Asp¹⁷⁵. They are Asp⁶¹, Ser¹¹⁰, Thr¹⁶³, Arg¹⁷⁴, Glu²⁷⁴, Asp²⁷⁹, His²⁸³, Asp³²¹, His³²², and His⁴¹⁴, which correspond to Asp⁷³, Ser¹²⁴, Thr¹⁷⁷, Arg¹⁸⁸, Glu³⁴⁴, Asp³⁴⁹, His³⁵³, Asp³⁹¹, His³⁹², and His⁴³⁴ of *E. coli* APase respectively. Of these, the serine and arginine residues form the catalytic site (phosphorylation site) and phosphate binding site respectively. Other residues form the three metal binding sites. These results suggest that SIB1 APase structurally and functionally resembles *E. coli* APase.

Asp¹⁷⁵ of *E. coli* APase, which is one of the constituents of the metal binding sites, is replaced by His (His¹⁶¹) in the SIB1 APase sequence. This residue is conserved as His, instead of Asp, in mammalian APases,³⁴ as well as the other APases listed in Fig. 1. It has been reported that Asp¹⁷⁵ of *E. coli* APase is responsible for its low specific activity as compared to those of mammalian APases,³⁴ and the substitution of this residue by His results in a great increase in enzymatic activity due to the increased binding affinity of Mg²⁺.³⁵ Another significant difference between the SIB1 and *E. coli* APase sequences is a deletion or an insertion of the sequence. The $\alpha 7$ helix and its flanking region (Cys¹⁹⁰–Gly²⁰⁸) and a long loop containing $\alpha 12$ helix (Val²⁸⁸–Cys³⁰⁸) of *E. coli* APase are deleted in SIB1 APase. On the other hand, a long loop consisting of 37 residues is inserted in SIB1 APase at a position corresponding to that between the $\beta 8$ and $\beta 9$ strands of *E. coli* APase, but the other APase sequences have similar deletions and insertions when compared to the *E. coli* APase sequence. Hence, these differences are attributable to the unique *E. coli* APase sequence.

Biochemical properties of the recombinant protein

We used *E. coli* AC109(DE3) as a host for overproduction of SIB1 APase to avoid contamination of host-derived APase. The recombinant protein was purified from the periplasmic space of the *E. coli* cells to give a single band on SDS–PAGE (Fig. 2). The amount of purified protein obtained from 1-liter culture was roughly 1.0 mg. The N-terminal amino acid sequence of the purified protein was found to be Asp–Glu–Val–Ile–Met–Pro–Pro–Ala, indicating that the N-terminal signal peptide (Met¹–Ala²⁷) was removed when the recombinant protein was secreted across the cytoplasmic membrane into the periplasmic space of the *E. coli* cells.

The molecular mass of the purified protein was estimated to be 44 kDa by SDS–PAGE and 100 kDa by gel filtration column chromatography (data not shown), suggesting that SIB1 APase exists as a dimer, like

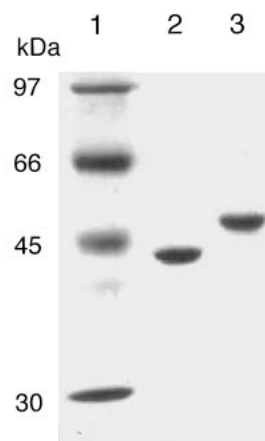


Fig. 2. SDS–PAGE of Purified SIB1 APase.

The samples were subjected to 12% SDS–PAGE and stained with Coomassie Brilliant Blue. Lane 1, a low molecular weight marker kit (Amersham Biosciences); lane 2, SIB1 APase; lane 3, *E. coli* APase. Numbers along the gel represent the molecular masses of individual standard proteins.

E. coli APase. The molecular mass of the protein estimated by SDS–PAGE was slightly smaller than but comparable to that (46,100) calculated from the amino acid sequence (Asp²⁸–Pro⁴⁵⁵). The far- and near-UV CD spectra of SIB1 APase were similar to those of *E. coli* APase, except for the height of a broad peak around 280 nm (Fig. 3). These results suggest that local conformations around the aromatic residues of SIB1 APase are slightly different from those of *E. coli* APase, but that its overall main chain fold is similar to that of *E. coli* APase.

Enzymatic activity

The optimum condition for enzymatic activity of SIB1 APase, except for the optimum temperature, was determined at 20 °C using *p*NPP as a substrate. The pH dependence of activity was analyzed using 50 mM Tris–HCl (pH 7.6–8.8) and 50 mM Gly–NaOH (pH 7.7–12.4). SIB1 APase exhibited the highest activity (a specific activity of 950 units/mg) at pH 10.5 (Fig. 4). Analysis of the dependence of activity on the concentration of Gly–NaOH (pH 10.5) indicated that it exhibited the highest activity at 50 mM (data not shown). It exhibited roughly 30% of maximal activity at 20 or 200 mM. Analyses of the dependence of activity on the NaCl and KCl concentrations indicated that the enzyme exhibited the highest activity in the presence of 100 mM KCl (data not shown). The enzyme exhibited roughly 50% of maximal activity in the absence of salt or the presence of 200 mM KCl. NaCl was rather inhibitory and enzymatic activity was reduced by 30% in the presence of 200 mM NaCl as compared to that determined in the absence of salt. Analyses for the dependence of activity on metal ions indicated that SIB1 APase activity was greatly enhanced by the simultaneous addition of Mg²⁺ and Zn²⁺ (Fig. 5). The enzyme gave a maximum specific activity of 950 u/mg in the presence of 20 mM MgCl₂

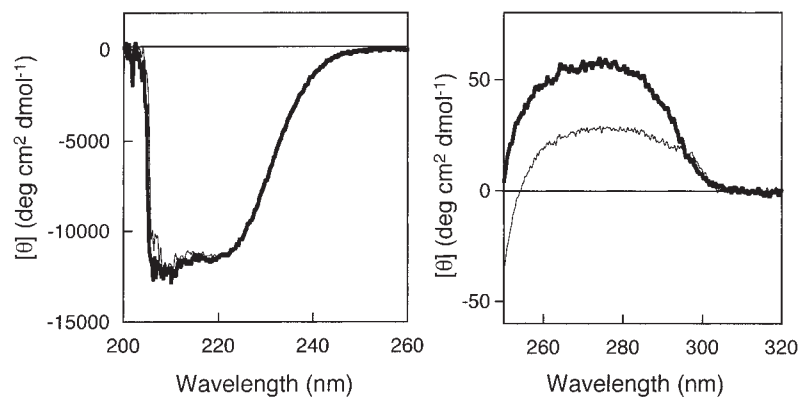


Fig. 3. CD Spectra.

The far-UV (left) and near-UV (right) CD spectra of SIB1 APase (thick line) and *E. coli* APase (thin line) are shown. These spectra were measured as described in "Materials and Methods".

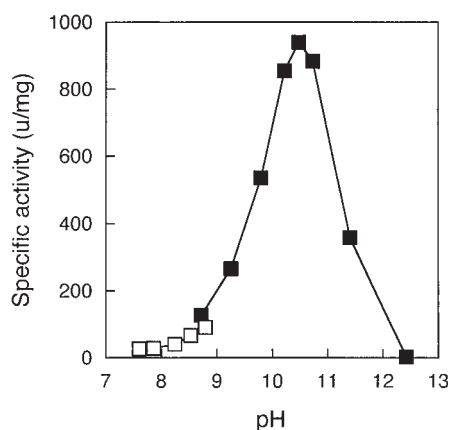


Fig. 4. pH Dependence of the Activity of SIB1 APase.

The enzymatic activity was determined at 20 °C in 50 mM Tris-HCl (pH 7.6–8.8) (□) or 50 mM Gly-NaOH (pH 7.7–12.4) (■) containing 20 mM MgCl₂, 5 mM ZnCl₂, 100 mM KCl, and 5 mM pNPP, as described in "Materials and Methods".

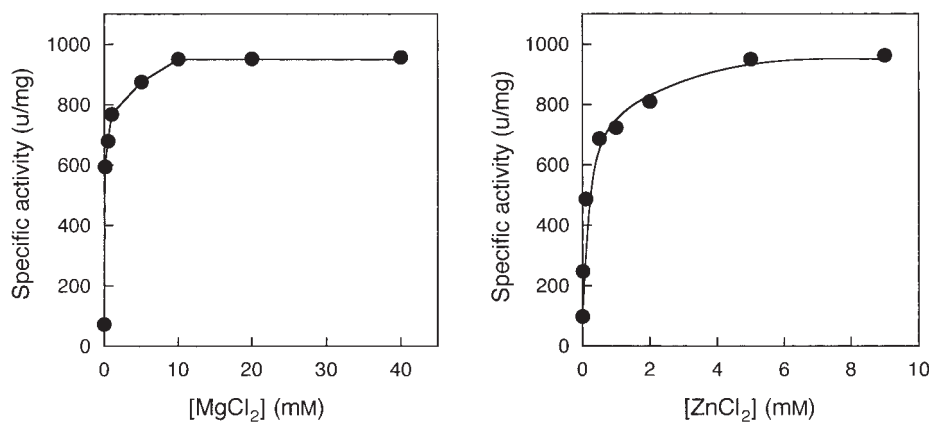


Fig. 5. Dependence of SIB1 APase Activity on Divalent Metal Ion Concentrations.

The enzymatic activities of SIB1 APase were determined at 20 °C in 50 mM Gly-NaOH (pH 10.5) containing 5 mM ZnCl₂, 100 mM KCl, 5 mM pNPP, and various concentrations of MgCl₂ (left), or the same buffer containing 20 mM MgCl₂, 100 mM KCl, 5 mM pNPP, and various concentrations of ZnCl₂ (right), as described in "Materials and Methods".

and 5 mM ZnCl₂. It exhibited roughly 5% of maximal activity in the absence of these metal ions (a specific activity of 50 u/mg). This activity increased only by at most twice when these metal ions were individually added to the reaction mixture. Other metal ions, such as Mn²⁺, Fe²⁺, Co²⁺, Ni²⁺, Ba²⁺, Cu²⁺, Ca²⁺, and Sr²⁺, did not significantly increase SIB1 APase activity.

The temperature dependence of SIB1 APase activity was analyzed by measuring the activity in 50 mM Gly-NaOH (pH 10.5) containing 100 mM KCl, 20 mM MgCl₂, 5 mM ZnCl₂, and 5 mM pNPP at various temperatures and comparing it with that of *E. coli* APase activity. The amounts of pNP accumulated on 10 min incubation with the enzyme were compared. SIB1 APase most effectively hydrolyzed the substrate at 50 °C, whereas *E. coli* APase did so at 80 °C (Fig. 6). Thus the optimum temperature for SIB1 APase activity was apparently shifted downward by roughly 30 °C as compared to that for *E. coli* APase activity. The specific activities of SIB1 and *E. coli* APases at various temperatures are summarized in Table 1. The specific activity

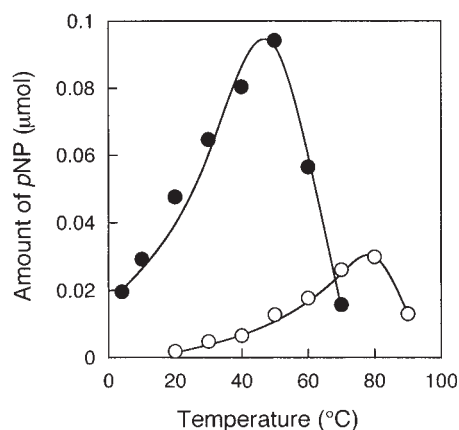


Fig. 6. Temperature Dependence of the Activities of SIB1 and *E. coli* APases.

The enzymatic reaction was carried out at the temperatures indicated in 1 ml of the reaction mixture for 10 min, and the amount of pNP accumulated upon reaction was plotted against the temperature. The reaction mixture contained 50 mM Gly–NaOH (pH 10.5), 20 mM MgCl₂, 5 mM ZnCl₂, 100 mM KCl, 5 mM pNPP, and 5 ng of SIB1 APase (●), or 0.1 M CAPS–NaOH (pH 10.0), 0.4 M NaCl, 10 mM MgCl₂, 5 mM pNPP, and 5 ng of *E. coli* APase (○).

Table 1. Specific Activities at Various Temperatures

Enzyme	Temperature (°C)	Specific activity (units/mg)
SIB1 APase	4	392
	20	952
	50	1880
<i>E. coli</i> APase	20	37.7
	50	257
	80	600

The specific activities of the enzymes were determined using pNPP as a substrate, as described in "Materials and Methods". The substrate concentration was 5 mM. The reaction was carried out for 10 min at the temperatures indicated. Each experiment was carried out in duplicate and the average value is shown. Errors from the average values were within 6%.

of SIB1 APase at 50 °C was 3.1 fold higher than that of *E. coli* APase at 80 °C, suggesting that the maximum hydrolysis rate of SIB1 APase is higher than that of *E. coli* APase. In addition, the specific activities of SIB1 APase were 7.3 and 25.3 fold higher than those of *E. coli* APase at 50 and 20 °C respectively, indicating that SIB1 APase adapts well to low temperatures as compared to *E. coli* APase. The specific activities of SIB1 and *E. coli* APases were not determined at temperatures higher than 55 and 85 °C respectively, because the amounts of pNP did not increase in proportion to increases in reaction time for 10 min at these temperatures. These enzymes probably are not fully stable for 10 min at these temperatures.

Stability

The stability of SIB1 APase against thermal inactivation is compared with that of *E. coli* APase in Fig. 7. *E. coli* APase was very stable and lost activity with a

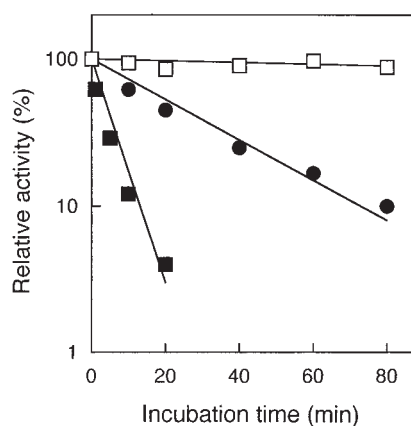


Fig. 7. Thermal Stability of SIB1 and *E. coli* APases.

SIB1 APase was incubated at 60 °C (●) and 70 °C (■), and *E. coli* APase was incubated at 80 °C (□) in 50 mM Tris–HCl (pH 8.0) containing 5 mM MgCl₂. At appropriate intervals, an aliquot was withdrawn and determined for residual activity at 40 °C for SIB1 APase and 60 °C for *E. coli* APase, as described in "Materials and Methods".

half-life of >6 h at 80 °C, whereas SIB1 APase lost activity with half-lives of 22 min at 60 °C and 3.9 min at 70 °C. Thus SIB1 APase was considerably less stable than *E. coli* APase.

Discussion

Cold-adaptation mechanism

In this study, the activity and stability of recombinant SIB1 APase were compared with those of *E. coli* APase, which represents mesophilic APases. Because of its decreased stability, downward shift of the optimum temperature for activity, and increased activity at low temperatures, SIB1 APase can be defined as a cold-adapted enzyme. Like other cold-adapted enzymes,^{1–3,6)} SIB1 APase might acquire a conformational flexibility at low temperatures at a cost in protein stability. An increase in the conformational flexibility of enzymes results in a better accommodation of the substrate and enhancement of the fast conformational changes required for catalysis. To understand the structural basis for the cold-adaptation of enzymes, the crystal structures of several cold-adapted enzymes have been determined and compared with those of mesophilic and thermophilic counterparts.⁴⁾ These studies indicate that all structural factors responsible for protein stabilization are attenuated in both strength and number in cold-adapted enzymes.

When the amino acid sequence of SIB1 APase (mature form) is compared with that of *E. coli* APase, the content of the charged residues (Lys+Arg+Glu+Asp) of SIB1 APase (17.5%) is significantly lower than that of *E. coli* APase (20.7%). Computational analysis has suggested that the decrease in content of charged residues causes protein destabilization.³⁶⁾ Charged residues are usually located at the surface of

protein molecules and stabilize proteins by electrostatic interactions. Therefore, a decrease in electrostatic interactions may be one of the factors responsible for the cold-adaptation of SIB1 APase. In addition, SIB1 APase does not contain cysteine residues, while *E. coli* APase contains four of them. Because these cysteine residues form two disulfide bonds¹⁰⁾ and disulfide bonds usually stabilize proteins,³⁷⁾ a decrease in the number of disulfide bonds might be another factor responsible for cold-adaptation of SIB1 APase. It has been reported that protein stability decreases as the content of proline residues decreases.³⁸⁾ Introduction of proline residues into loop regions has been reported to increase protein stability by decreasing the entropy of the unfolded state of proteins.³⁹⁾ However, the content of proline residues of SIB1 APase (4.9%) is comparable to that of *E. coli* APase (4.7%).

A comparison of the amino acid sequence of SIB1 APase with those of other APases, shown in Fig. 1, indicates that decreases in content of charged residues and number of disulfide bonds are not characteristic features of cold-adapted APases. The content of charged residues of TAB5 APase (20.7%) is comparable to that of *E. coli* APase, and mesophilic MR-1 APase and *B. subtilis* APase III do not contain cysteine residues. However, the content of proline residues of TAB5 APase (2.3%) is considerably lower than that of SIB1 or *E. coli* APase. These results suggest that each protein adopts its own strategy for cold-adaptation. It is difficult to construct a three-dimensional model for the SIB1 APase structure, because *E. coli* APase is the only APase for which the crystal structure is available,¹⁰⁾ and the SIB1 APase sequence has large deletions and insertions at several positions as compared to the *E. coli* APase sequence (Fig. 1). Therefore, it is necessary to determine the crystal structure of SIB1 APase to understand its cold-adaptation mechanism. Availability of the overproduction system for SIB1 APase will facilitate crystallographic studies of it.

It has been suggested that deletion of the $\alpha 7$ helix is responsible for the cold-adaptation of SCAPase.¹⁹⁾ However, this helix is deleted not only in psychrophilic APases but also in mesophilic and thermophilic ones (Fig. 1). Likewise, a large peptide containing the $\alpha 12$ helix of *E. coli* APase is deleted and a large peptide is inserted at the position corresponding to that between the $\beta 8$ and $\beta 9$ strands of *E. coli* APase not only in psychrophilic APases but also in mesophilic and thermophilic ones (Fig. 1). These results suggest that these deletions and insertions are not responsible for the cold-adaptation of SIB1 APase.

According to the crystal structure of *E. coli* APase, two monomers interact tightly to form a homodimer.¹⁰⁾ The amino acid residues, which are located mainly in the N-terminal region ($\alpha 1$ – $\alpha 3$ helices and $\beta 2$ strand) and the C-terminal region ($\beta 9$ and $\beta 10$ strands and $\alpha 16$ helix), form the interface of a homodimer. When the amino acid sequence of SIB1 APase is compared with

that of *E. coli* APase, these interface residues are more variable than those, which are located in the core region (the $\alpha 4$, $\alpha 6$, $\alpha 13$, and $\alpha 14$ helices and the $\beta 1$, $\beta 3$, $\beta 4$, $\beta 6$, $\beta 7$, $\beta 8$ strands) (Fig. 1). This high frequency of amino acid substitutions in the interface may account for the low stability and high flexibility of SIB1 APase as compared to *E. coli* APase, because homodimeric proteins are expected to be destabilized by reducing their monomer-monomer interactions. However, the area of the monomer-monomer interface is too large (20% of the surface area) to identify the amino acid substitutions responsible for destabilization of SIB1 APase by simply comparing the amino acid sequences.

Possible application

APases have been used to remove 5'-terminal phosphate groups from DNA fragments. *E. coli* APase (BAP) has been most widely used for this purpose. However APases should be completely inactivated or eliminated to ligate or phosphorylate the resultant DNA fragments. Because BAP is highly thermostable, phenol/chloroform treatment and ethanol precipitation are necessary to inactivate it. Recombinant SIB1 APase is therefore expected to be more useful than BAP, because a simple heat treatment inactivates the enzyme and might allow subsequent ligation or phosphorylation of DNA fragments successively. Recombinant SIB1 APase has been shown to remove 5'-terminal phosphate groups effectively not only from DNA fragments with 5'-overhangs but also from those with 3'-overhangs and blunt ends (H. Nagahora, personal communication). Note that shrimp APase has often been used as a heat labile APase. SIB1 APase is superior to this APase, however, because of ease of production and purification.

The gene encoding APase (SCAPase) has been cloned from the same genus *Shewanella* and the recombinant protein has been overproduced in *E. coli* and characterized.¹⁹⁾ However, comparison of the amino acid sequence of SCAPase with those of SIB1 and MR-1 APases indicates that the SCAPase sequence lacks the N-terminal 40–50 residues, which contain the signal sequence required for secretion and the $\alpha 1$ helix (Fig. 1). Because the genus *Shewanella* is a Gram-negative bacterium like *E. coli* and APases from Gram-negative bacteria are secreted into the periplasmic space of the cells, SCAPase should have the N-terminal signal sequence. The sequence reported for SCAPase probably represents an incomplete one which lacks the N-terminal 40–50 residues. In addition, recombinant SCAPase is apparently more stable than recombinant SIB1 APase, because its optimum temperature for activity has been reported to be 70°C. However, the optimum temperature of natural SCAPase for activity has been reported to be 40°C. Because recombinant SCAPase was overproduced in the *E. coli* strain without the *phoA* mutation and the yield of the purified protein was extremely low (1.7 mg from 30 g of the cells), its activity may represent that of *E. coli* APase derived from the host strain.

Alternatively, deletion of the N-terminal sequence containing the $\alpha 1$ helix might cause an upward shift of the optimum temperature. TAB5 APase, another psychrophilic APase, has been reported to effectively remove 5'-terminal phosphate groups from DNA fragments.¹⁷⁾ However, recombinant TAB5 APase is recovered in membrane fractions and can be dissolved in multimeric form by detergent (0.1% Triton X-100).

Acknowledgments

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References

- 1) Feller, G., Narinx, E., Arpigny, J. L., Aittaleb, M., Baise, E., Genicot, S., and Gerday, C., Enzymes from psychrophilic organisms. *FEMS Microbiol. Rev.*, **18**, 189–202 (1996).
- 2) Marshall, C. J., Cold-adapted enzymes. *TIBTECH*, **15**, 359–364 (1997).
- 3) Russell, N. J., Toward a molecular understanding of cold activity of enzymes from psychrophiles. *Extremophiles*, **4**, 83–90 (2000).
- 4) Feller, G., and Gerday, C., Psychrophilic enzymes: hot topics in cold adaptation. *Nat. Rev. Microbiol.*, **1**, 200–208 (2003).
- 5) Gianese, G., Bossa, F., and Pascarella, S., Comparative structural analysis of psychrophilic and meso- and thermophilic enzymes. *Proteins*, **47**, 236–249 (2002).
- 6) Smalas, A. O., Leiros, H. K., Os, V., and Willassen, N. P., Cold adapted enzymes. *Biotechnol. Annu. Rev.*, **6**, 1–57 (2000).
- 7) Gerday, C., Aittaleb, M., Bentahir, M., Chessa, J. P., Claverie, P., Collins, T., D'Amico, S., Dumont, J., Garsoux, G., Georgette, D., Hoyoux, A., Lonhienne, T., Meuwis, M. A., and Feller, G., Cold-adapted enzymes: from fundamentals to biotechnology. *TIBTECH*, **18**, 103–107 (2000).
- 8) Cavicchioli, R., Siddiqui, K. S., Andrews, D., and Sowers, K. R., Low-temperature extremophiles and their applications. *Curr. Opin. Biotechnol.*, **13**, 253–261 (2002).
- 9) McComb, R. B., Bowers, G. N., and Posen, S., "Alkaline Phosphatase", Plenum Press, New York (1979).
- 10) Kim, E. E., and Wyckoff, H. W., Reaction mechanism of alkaline phosphatase based on crystal structures. Two-metal ion catalysis. *J. Mol. Biol.*, **218**, 449–464 (1991).
- 11) Ma, L., and Kantrowitz, E. R., Kinetic and X-ray structural studies of a mutant *Escherichia coli* alkaline phosphatase (His-412→Gln) at one of the zinc binding sites. *Biochemistry*, **35**, 2394–2402 (1996).
- 12) Tibbitts, T. T., Murphy, J. E., and Kantrowitz, E. R., Kinetic and structural consequences of replacing the aspartate bridge by asparagine in the catalytic metal triad of *Escherichia coli* alkaline phosphatase. *J. Mol. Biol.*, **257**, 700–715 (1996).
- 13) Murphy, J. E., Stec, B., Ma, L., and Kantrowitz, E. R., Trapping and visualization of a covalent enzyme-phosphate intermediate. *Nat. Struct. Biol.*, **4**, 618–622 (1997).
- 14) Holtz, K. M., Stec, B., and Kantrowitz, E. R., A model of the transition state in the alkaline phosphatase reaction. *J. Biol. Chem.*, **274**, 8351–8354 (1999).
- 15) Wojciechowski, C. L., and Kantrowitz, E. R., Altering of the metal specificity of *Escherichia coli* alkaline phosphatase. *J. Biol. Chem.*, **277**, 50476–50481 (2002).
- 16) Orhanovic, S., and Pavela-Vrancic, M., Dimer asymmetry and the catalytic cycle of alkaline phosphatase from *Escherichia coli*. *Eur. J. Biochem.*, **270**, 4356–4364 (2003).
- 17) Rina, M., Pozidis, C., Mavromatis, K., Tzanodaskalaki, M., Kokkinidis, M., and Bouriotis, V., Alkaline phosphatase from the Antarctic strain TAB5. Properties and psychrophilic adaptations. *Eur. J. Biochem.*, **267**, 1230–1238 (2000).
- 18) Asgeirsson, B., and Andresson, O. S., Primary structure of cold-adapted alkaline phosphatase from a *Vibrio* sp. as deduced from the nucleotide gene sequence. *Biochim. Biophys. Acta*, **1549**, 99–111 (2001).
- 19) Murakawa, T., Yamagata, H., Tsuruta, H., and Aizono, Y., Cloning of cold-active alkaline phosphatase gene of a psychrophile, *Shewanella* sp., and expression of the recombinant enzyme. *Biosci. Biotechnol. Biochem.*, **66**, 754–761 (2002).
- 20) Ishida, Y., Tsuruta, H., Tsuneta, S. T., Uno, T., Watanabe, K., and Aizono, Y., Characteristics of psychrophilic alkaline phosphatase. *Biosci. Biotechnol. Biochem.*, **62**, 2246–2250 (1998).
- 21) Kato, T., Haruki, M., Imanaka, T., Morikawa, M., and Kanaya, S., Isolation and characterization of psychrotrophic bacteria from oil reservoir water and oil sands. *Appl. Microbiol. Biotech.*, **55**, 794–800 (2001).
- 22) Ohtani, N., Haruki, M., Morikawa, M., and Kanaya, S., Heat labile ribonuclease HI from a psychrotrophic bacterium: gene cloning, characterization, and site-directed mutagenesis. *Protein Eng.*, **14**, 975–982 (2001).
- 23) Suzuki, Y., Haruki, M., Takano, K., Morikawa, M., and Kanaya, S., Possible involvement of a FKBP family protein from a psychrotrophic bacterium *Shewanella* sp. SIB1 in cold-adaptation. *Eur. J. Biochem.*, **271**, 1372–1381 (2004).
- 24) Elvin, C. M., Hardy, C. M., and Rosenberg, H., Pi exchange mediated by the GlpT-dependent sn-glycerol-3-phosphate transport system in *Escherichia coli*. *J. Bacteriol.*, **161**, 1054–1058 (1985).
- 25) Imanaka, T., Fujii, M., and Aiba, S., Isolation and characterization of antibiotic resistance plasmids from thermophilic bacilli and construction of deletion plasmids. *J. Bacteriol.*, **146**, 1091–1097 (1981).
- 26) Brockman, R. W., and Heppel, L. A., On the localization of alkaline phosphatase and cyclic phosphodiesterase in *Escherichia coli*. *Biochemistry*, **7**, 2554–2562 (1968).

- 27) Laemmli, U. K., Cleavage of structural proteins during the assembly of the head of bacteriophage T4. *Nature*, **227**, 680–685 (1970).
- 28) Goodwin, T. W., and Morton, R. A., The spectrophotometric determination of tyrosine and tryptophan in proteins. *Biochem. J.*, **40**, 628–632 (1946).
- 29) Muller, B. H., Lamoure, C., Le Du, M. H., Cattolico, L., Lajeunesse, E., Lemaitre, F., Pearson, A., Ducancel, F., Menez, A., and Boulain, J. C., Improving *Escherichia coli* alkaline phosphatase efficacy by additional mutations inside and outside the catalytic pocket. *Chembiochem.*, **2**, 517–523 (2001).
- 30) Heidelberg, J. F., Paulsen, I. T., Nelson, K. E., Gaidos, E. J., Nelson, W. C., Read, T. D., Eisen, J. A., Seshadri, R., Ward, N., Methe, B., Clayton, R. A., Meyer, T., Tsapin, A., Scott, J., Beanan, M., Brinkac, L., Daugherty, S., DeBoy, R. T., Dodson, R. J., Durkin, A. S., Haft, D. H., Kolonay, J. F., Madupu, R., Peterson, J. D., Umayam, L. A., White, O., Wolf, A. M., Vamathevan, J., Weidman, J., Impraim, M., Lee, K., Berry, K., Lee, C., Mueller, J., Khouri, H., Gill, J., Utterback, T. R., McDonald, L. A., Feldblyum, T. V., Smith, H. O., Venter, J. C., Nealon, K. H., and Fraser, C. M., Genome sequence of the dissimilatory metal ion-reducing bacterium *Shewanella oneidensis*. *Nat. Biotechnol.*, **20**, 1118–1123 (2002).
- 31) Chang, C. N., Kuang, W. J., and Chen, E. Y., Nucleotide sequence of the alkaline phosphatase gene of *Escherichia coli*. *Gene*, **44**, 121–125 (1986).
- 32) Hulett, F. M., Kim, E. E., Bookstein, C., Kapp, N. V., Edwards, C. W., and Wyckoff, H. W., *Bacillus subtilis* alkaline phosphatases III and IV. Cloning, sequencing, and comparisons of deduced amino acid sequence with *Escherichia coli* alkaline phosphatase three-dimensional structure. *J. Biol. Chem.*, **266**, 1077–1084 (1991).
- 33) Wojciechowski, C. L., Cardia, J. P., and Kantrowitz, E. R., Alkaline phosphatase from the hyperthermophilic bacterium *T. maritima* requires cobalt for activity. *Protein Sci.*, **11**, 903–911 (2002).
- 34) Murphy, J. E., and Kantrowitz, E. R., Why are mammalian alkaline phosphatases much more active than bacterial alkaline phosphatases? *Mol. Microbiol.*, **12**, 351–357 (1994).
- 35) Murphy, J. E., Xu, X., and Kantrowitz, E. R., Conversion of a magnesium binding site into a zinc binding site by a single amino acid substitution in *Escherichia coli* alkaline phosphatase. *J. Biol. Chem.*, **268**, 21497–21500 (1993).
- 36) Nakashima, H., Fukuchi, S., and Nishikawa, K., Compositional changes in RNA, DNA and proteins for bacterial adaptation to higher and lower temperatures. *J. Biochem. (Tokyo)*, **133**, 507–513 (2003).
- 37) Thornton, J. M., Disulphide bridges in globular proteins. *J. Mol. Biol.*, **151**, 261–287 (1981).
- 38) Watanabe, K., Chishiro, K., Kitamura, K., and Suzuki, Y., Proline residues responsible for thermostability occur with high frequency in the loop regions of an extremely thermostable oligo-1,6-glucosidase from *Bacillus thermoglucosidasius* KP1006. *J. Biol. Chem.*, **266**, 24287–24294 (1991).
- 39) Matthews, B. W., Nicholson, H., and Becktel, W. J., Enhanced protein thermostability from site-directed mutations that decrease the entropy of unfolding. *Proc. Natl. Acad. Sci. U.S.A.*, **84**, 6663–6667 (1987).